

TITLE OF THE INVENTION

8
26
METHODS OF SCREENING FOR
~~METHODS OF SEARCHING FOR SOLID FORMS AND~~
~~POSSIBLE SOLID FORMS~~
~~SCREENING A SAMPLE ACCORDING TO ITS FORMS~~

FIELD OF THE INVENTION

5 The present methods relate to searching for possible forms of a sample and include solidifying the sample in at least one receptacle defining a capillary space. The present methods also relate to screening a sample according to its forms and include solidifying the sample in a plurality of receptacles, and at least one receptacle defines a capillary space. The form of
10 the sample refers to its arrangement at the molecular or atomic level in the solid. The forms generated by solidification comprise solid forms and semisolid forms. The generated forms are analyzed and classified, such as by x-ray
15 diffraction patterns. The present methods increase the likelihood of generating all or a high percentage of possible forms.

BACKGROUND OF THE INVENTION

20 In the chemical field, the unpredictability and variability of compounds, mixtures, and processes are well established. Certain chemical compounds or mixtures may have utility for numerous different applications, including vital

biological applications, yet a slight change in those compounds or mixtures, even with respect to a single atom, may reduce or eliminate their utility for their beneficial purpose. Similarly,
5 certain chemical processes may have significantly better or worse performance based upon seemingly minor differences.

In the pharmaceutical field, a great deal of time, effort and expense is spent on the
10 identification of particular compounds and mixtures that will have beneficial effect. Furthermore, exhaustive research must be done as to whether such compounds and mixtures will have harmful effects. Once again, even slight
15 differences in chemical composition or structure may yield significant differences in biological activity. Thus, researchers frequently test many different compounds and mixtures for biological activity and other effects as well as testing
20 different processes and conditions for the preparation of such chemical compounds and mixtures.

The process of thorough analysis of different chemical compounds, elements, mixtures,
25 processes, or structures is commonly referred to as screening. Screening may be a function of time and effort, with the quality or results of screening being a function of the number of samples prepared and/or analyzed as well as the
30 quality of preparation and/or analysis underlying those samples. Screening plays a vital role in the pharmaceutical field, as the most

advantageous formulation of a biologically active compound or mixture is frequently found through successful screening processes.

However, screening processes can require
5 significant amounts of time, effort and resources. There is a continuous need for improved screening processes having increased reliability and efficiency.

Processes have been used for screening
10 chemical compounds according to their form. When a compound has different solid or crystalline forms, the different forms are frequently referred to as polymorphs of the compound. A "polymorphic" compound as used herein means a
15 compound having more than one solid form. For example, a polymorphic compound may have different forms of its crystalline structure, or different forms based upon hydration, or it may have a crystalline form and an amorphous form.
20 In the past, screening processes have not identified with sufficient consistency and reliability a high percentage of possible solid and semisolid forms.

The form of a compound or mixture may have
25 an impact on biological activity. The same chemical compound may exhibit different properties depending upon which form (such as amorphous or crystalline or semisolid) that compound is in. A "semisolid" form is used herein
30 to indicate materials like waxes, suspensions, gels, creams, and ointments. The term "solid form" herein includes semisolid forms.

Furthermore, a chemical compound may exist in different solid forms, and those different solid forms may also exhibit different properties. As a result, different solid forms, including
5 different crystalline forms, of a chemical compound may have greater or lesser efficacy for a particular application. The identification of an optimal solid form is important in the pharmaceutical field, as well as in other fields
10 including nutraceuticals, agricultural chemicals, dyes, explosives, polymer additives, lubricant additives, photographic chemicals, and structural and electronic materials. The new methods described herein may be useful in any of these
15 fields as well as others where solid materials are used.

A chemical compound or mixture may be amorphous, meaning that it is not characterized by a regular arrangement of molecules.
20 Alternatively (or even to a limited extent within a mostly amorphous form), a compound or mixture may be arranged in a crystalline state, where the molecules exist in fixed conformations and are arranged in a regular way. The same compound or
25 mixture may exhibit different properties depending upon which solid form that compound or mixture is in.

It is important in the pharmaceutical field as well as other fields to find the form of a
30 chemical compound that exhibits appropriate physical and chemical properties. One form may be more stable or have other properties that make

it preferable over other forms. One form of a chemical composition may have better bioavailability, solubility, or adsorption characteristics or in other ways be more suitable for delivery of therapeutic doses than other forms. As part of a screening method, it may be advisable to evaluate different salts of a chemical compound (or more precisely, different salt compounds of a given biologically active ion). It is frequently desirable within a screening process to generate, or at least search for, a high percentage of the possible solid forms of a compound or mixture. Past attempts to generate a variety of solid forms involved flash evaporations, cooling under different conditions and/or the addition of seeds of solid material. However, some materials strongly resist the generation of new solid forms.

One or more solid forms may be generated by crystallization of the sample. Among the phenomena in crystallization are nucleation and growth. Crystal nucleation is the formation of an ordered solid phase from liquids, supersaturated solutions, saturated vapors, or amorphous phases.

Nucleation may be achieved by homogeneous or heterogeneous mechanisms. In heterogeneous mechanisms, some solid particle is present to provide a catalytic effect and reduce the energy barrier to formation of a new phase. Crystals may originate on a minute trace of a foreign substance (either impurities or container walls) acting as a nucleation site. Since nucleation may

set the character of the crystallization process, the identity of the foreign substance is an important parameter. The presence of "seeds" of other crystalline compounds in a crystallization
5 environment can be beneficial, detrimental, or both, but in any event, must be considered. Growth is the enlargement of crystals caused by deposition of molecules on an existing surface. In homogeneous mechanisms, it has been theorized
10 by others that nucleation is achieved spontaneously with the solution comprising the solute to be crystallized in solvent typically by evaporation, temperature reduction, or addition of antisolvent.

15 Typically, a solid to be crystallized is present in a solution at, above, or below its saturation point at a given temperature. Crystallization is initiated or facilitated by removing solvent, changing temperature, and/or
20 adding an antisolvent. The solvent may be removed by evaporation or other means. Eventually the solution reaches a point where crystals will grow.

A specific chemical substance may
25 crystallize into different forms or transition from one polymorph form, pseudopolymorph form, or amorphous form to another form. This crystallization into a different form or transition into a different form may be
30 accompanied by other physical or chemical changes. For example, novobiocin has at least two different forms: an amorphous form and a

crystalline form. Dog plasma levels of novobiocin vary depending on which form of novobiocin is administered. In one study, two hours after the amorphous form of the drug was administered, the concentration of novobiocin was 29.3 mg/mL. In contrast, when crystalline novobiocin was administered, there was no drug detectable in the dog plasma two hours after the drug was administered. In another example, furosemide has two different crystalline forms, and furosemide solubility in aqueous buffer at pH 3.2 varied depending on which polymorph was studied. After three hours, Form I and Form II had solubilities of approximately 0.025 mg/mL. Under the same conditions and dissolution time, the DMF and dioxane solvates of furosemide had solubilities of approximately 0.035, and Form III had a solubility of approximately 0.045 g/mL.

It is known to generate crystalline samples in capillary tubes. For example, U.S. Patent No. 5,997,636 discusses a method for growing crystals within a capillary tube. As another example, D. Amaro-González et al., "Gas Antisolvent Crystallization Of Organic Salts From Aqueous Solution", Journal Of Supercritical Fluids, 17 (2000) 249-258, discloses results of crystallization of lobenzarit, including crystallizations in capillaries. Lobenzarit is an anti-arthritis agent. Amaro-González et al. state that particle size and agglomeration varied depending on the size of the capillary, that it is shown that the size distribution and particle

shape can be controlled using different capillary diameters, and that it is possible to obtain individual crystals without agglomeration.

Neither reference discloses that different
5 forms (meaning different arrangements on the molecular or atomic level) were produced, nor does either reference suggest a new method for searching for possible forms or screening a sample according to its form. A different
10 particle size or shape does not necessarily mean there is a different crystal form since a solid form can crystallize into many different shapes. For example, snowflakes may comprise a single crystal form having many different crystal
15 shapes.

It is also known to subject samples within capillary tubes to various spectroscopic analyses, including diffraction analysis such as x-ray diffraction analysis. However, in such
20 instances, it has been the common practice to prepare a solid sample outside the capillary tube before it is placed in the capillary tube for analysis.

There are several factors that discourage
25 the use of capillary tubes for solidifying compounds or mixtures. One factor is that capillary tubes are more difficult to work with than other containers. Another factor is that there has been no general recognition that the
30 use of capillary spaces may affect reactions or lead to compositional or chemical differences. Thus, since it was believed that the same forms

and reactions could be done in other containers,
it is believed that capillary tubes have not been
used as an integral part of a screening process
or to search for and generate solid and semisolid
5 forms.

There is a need for improved screening
methods that identify all or a high percentage of
possible forms of a compound or mixture. There
is a need for improved methods of searching for
10 the possible forms of a sample.

SUMMARY OF THE INVENTION

As one aspect, an improved method of
searching for possible forms of a sample is
15 provided. The method comprises the steps of
disposing the sample on one or more receptacles,
where at least one of the receptacles defines a
capillary space, and the sample is disposed
within the capillary space. The method next
20 comprises solidifying the sample in or on the
receptacles to generate at least one form,
wherein the generated form(s) is a solid or
semisolid. The form(s) is then analyzed and
classified, such as by classification according
25 to what form it is.

As another aspect, an improved method of
screening a sample according to its form is
provided. This method is especially useful for
screening a sample comprising a compound or a
30 mixture having biological activity in at least
one form of the compound or mixture. The
screening method comprises the steps of disposing

the sample on a plurality of receptacles, where at least one of the receptacles defines a capillary space, and the sample is disposed within the capillary space. The method next
5 comprises solidifying the sample in or on the receptacles to generate at least one form, wherein at least one form is a solid or semisolid. The method further comprises analyzing at least one form in a manner wherein
10 the analytical result is indicative of the generated form(s), and classifying the generated form(s), such as by form type or according to analytical result.

The screening method may be particularly
15 useful where the compound or mixture has at least one form having biological application and it is desirable to determine if other forms are possible. The present methods may comprise generating at least one other form of the
20 compound or mixture.

The sample may comprise a known polymorphic compound or comprise at least one material that is not recognized as a polymorphic compound. The sample may consist essentially of a solution of
25 one compound, or may comprise a mixture of compounds.

Preferably, the present methods include disposing the sample on a plurality of receptacles, including at least two different
30 types of receptacles. For example, one portion of a sample may be disposed in a capillary tube that defines a capillary space and another

portion of the sample may be disposed on a glass slide that does not define a capillary space. The sample may be prepared in a single batch or in multiple batches. After the portions have solidified, the form disposed in the capillary tube and the form disposed on the slide may be analyzed, classified and compared.

A preferred receptacle defining a capillary space is a capillary tube, and others include a well plate, a block and a sheet with holes or pores of appropriate size and shape.

The present methods may further comprise the step of comparing the generated form to a known form. In many cases, the generating step may produce at least one different form of the sample.

At least some of the receptacles may be subjected to substantially constant motion during the generating step. For example, a capillary tube may be rotated along its longitudinal axis during the generating step or subjected to centrifuging during the generating step. Centrifuging can be sufficient to concentrate the solid or semisolid at one end of a capillary tube and to facilitate in-situ analysis of the generated forms. Also, variations in centrifuging may provide environmental variation, which is desired in a screening method. Centrifuging may move the sample to the bottom of the receptacle when one end of the receptacle is closed. Centrifuging may be performed at a

pressure lower than ambient pressure, or under vacuum.

In the present methods, the sample may comprise a compound comprising a biologically active ion or one or more different salts of the compound. A second analyzing step may be performed on generated forms, where the second analyzing step provides data indicative of biological activity or bioavailability.

In the present methods, the generated forms may be analyzed by any suitable means, such as methods selected from the group consisting of visual analysis, microscopic analysis, thermal analysis, diffraction analysis, and spectroscopic analysis. Preferred methods of analysis include Raman spectroscopic analysis and x-ray diffraction analysis, more preferably using synchrotron radiation as the radiation source for the analysis. The analysis may determine differences in arrangement of molecules in the solid or determine one or more other characteristics that directly or indirectly reflect the form.

In the present methods, the step of analyzing the generated form may comprise analyzing the form without removing it from the receptacle in which it was generated. Thus, the present methods are useful for in situ analysis of generated forms. The use of capillary tubes as receptacles can facilitate such in situ analysis.

It may be advantageous to place the sample in at least five receptacles defining capillary spaces, alternatively at least 100 receptacles defining capillary spaces. In some embodiments, a sample is placed in several sets of numerous capillary tubes (for example, from 5 to 2000 capillary tubes, alternatively 5 to 100 capillary tubes), and the different sets are subjected to different methods or conditions of solidification.

The solidifying step may comprise crystallizing the sample, or may be selected from the group consisting of solvent evaporation, cooling, anti-solvent addition, gel diffusion, and thin-layer deposition.

A supersaturated solution of the sample can be formed before, during, or after the sample is disposed on the receptacle(s).

The generating step preferably comprises crystallizing the sample, or alternatively is selected from the group of methods consisting of solvent evaporation, cooling, anti-solvent addition, gel diffusion, and thin-layer deposition (with or without subsequent measures to quickly remove residual solvent, including air of various temperatures forced through the capillaries).

The receptacle that defines a capillary space can be a capillary tube or appropriately sized multi-well plate. Alternatively, the receptacle that defines a capillary space may be a block or a sheet made of polymer, glass, or

other material, which has holes or pores of a suitable shape and dimensions. Alternatively, some receptacles need not define a capillary space; indeed, it is considered preferable to
5 employ different kinds of receptacles for generating solid and/or semisolid forms of a given sample. Additional receptacles may include a glass slide or a conveyer surface in addition to the receptacle(s) defining capillary
10 spaces.

DETAILED DESCRIPTION OF PREFERRED EMBODIMENTS

The use of receptacles that define capillary spaces is an improvement over more labor-
15 intensive methods of generating solid forms and enables one to obtain a high percentage of possible solid and semisolid forms. Another advantage of such receptacles is that smaller amounts of the compound or mixture are used. A
20 compound is a substance composed of atoms or ions in chemical combination. A compound usually is composed of two or more elements, though as used in accordance with the present methods, a compound may be composed of one element.

25 A "polymorph" as used herein means a compound or mixture having more than one solid or semisolid form. The "form" of a compound or mixture refers to the arrangement of molecules in the solid. A "semisolid" form is used herein
30 to indicate materials like waxes, suspensions, gels, creams, and ointments. The term "solid form" herein includes semisolid forms.

"Capillary space" is defined herein to mean a space having walls separated by from about 0.1 mm to about 30 mm, preferably from about 0.5 mm to about 5 mm, more preferably from about 0.5 mm to about 2.5 mm, in at least one dimension. A capillary tube having an inner diameter from about 0.5 mm to about 2.5 mm, is a preferred receptacle that defines a capillary space in the interior of the capillary tube. It is preferred that the capillary tubes are circular in their interior shapes.

As used herein, the generation of solid and semisolid forms includes any suitable technique for solidification including but not limited to crystallization. Indeed, the forms which may be sought or generated may include amorphous forms, mixtures of amorphous forms, eutectic mixtures, mixed crystal forms, solid solutions, co-crystals, and other forms.

In certain embodiments of the present methods, solid samples are generated in receptacles through a suitable means of solidification. Typically, a solution containing a compound or mixture to be solidified and a solvent is placed in a receptacle defining a capillary space, such as a capillary tube. The compound or mixture can be present in a solution below, at or above its saturation point at a given temperature at the time it is placed in a capillary tube. Through evaporation, the use of an antisolvent, temperature variation, and/or other suitable means, the system reaches a point

where solidification begins. After a suitable amount of time, when solid or semisolid appears, the resulting sample is ready for analysis.

Any suitable crystallization technique may
5 be employed for obtaining crystals. For example, crystals may be obtained through cooling, heating, evaporation, addition of an antisolvent, reactive crystallization, and using supercritical fluids as solvents. Additionally, melt
10 crystallization techniques may be used to generate a solid form. Through such techniques, the use of a solvent can be avoided. In such techniques, formation of crystalline material is from a melt of the crystallizing species rather
15 than a solution. Additionally, the crystallization process may be done through sublimation techniques.

Crystallization may be performed as a seeded operation or an unseeded operation. In a seeded
20 operation, a selected quantity of seed crystals is included in the system. The characteristics of the seed crystals typically influence the characteristics of the crystals generated from the system. Crystallization may be performed by
25 heterogeneous or homogeneous mechanisms.

In other embodiments of the present methods, the form is generated other than by crystallization. The sample may be in the form of a melt that is then added to the capillary tube
30 and allowed to solidify in an amorphous form. Alternatively, the mechanism by which solidification is accomplished may include gel

diffusion methods, thin-layer deposition methods,
or other suitable methods. Other thermodynamic
and kinetic conditions may be employed to
solidify the compound or mixture. Cooling of a
5 saturated solution is a typical thermodynamic
condition. An addition of a solution of the
compound or mixture to an excess of cold anti-
solvent is a typical kinetic condition.

Any material capable of forming a solid or
10 semisolid may be used in the present methods. In
particular, the present methods are especially
suited for materials characterized by molecules
which are associated by non-bonded interactions
(e.g. van der Waals forces, hydrogen bonding, and
15 Columbic interaction).

The present methods may be advantageously
used with small organic drug molecules having
solubility of at least 1 mg/mL in ethanol at
ambient conditions. The present methods are also
20 contemplated for use with large organic molecules
and inorganic molecules. Examples of compounds
having more than one solid form include 5-methyl-
2-[(2-nitrophenyl)amino]-3-thiophenecarbonitrile
and 4-methyl-2-nitroacetanilide, each of which
25 may be different colors in connection with
different forms, and novobiocin and furosemide,
which are discussed above. This list cannot be
exhaustive as the present methods may provide
significant benefits for novel compounds and
30 mixtures whose identities, or at least whose
possible forms, are not yet identified.

The generation of a variety of forms is an important object of screening. A sufficient number of diverse processes and conditions should be employed to maximize the likelihood that a high percentage of possible solid forms of a chemical compound is generated. Samples should be generated under various thermodynamic and kinetic conditions.

It is preferable that the generation of solid and/or semisolid forms within the receptacles is carried out under a wide variety of conditions. For example, solids should be generated in the presence and absence of various solvents, as the solvent may play a role in the formation of certain forms.

As another example it is also preferable to prepare samples under different conditions of temperature and pressure, as different solid forms may be favored by different conditions.

The various forms generated may be identified by any suitable method, including but not limited to visual analysis (such as when different forms exhibit different colors), microscopic analysis including electron microscopy, thermal analysis such as determining the melting points, conducting diffraction analysis (such as x-ray diffraction analysis, electron diffraction analysis, neutron diffraction analysis, as well as others), conducting an infrared spectroscopic analysis, or conducting other spectroscopic analysis. Any appropriate analytical technique that is used to

differentiate structural, energetic, or performance characteristics may be used in connection with the present methods.

5 The classifying step may comprise classifying the generated form(s) according to any of the analytical results, such as appearance, solubility, or x-ray diffraction pattern.

10 In a preferred embodiment, a synchrotron may be used as the source of radiation for conducting diffraction analyses. A synchrotron is a type of particle accelerator, which emits high energy, focused radiation. Synchrotron radiation is the byproduct of circulating electrons or positrons
15 at speeds very close to the speed of light. Synchrotron radiation contains all the wavelengths of the electromagnetic spectrum and comprises the most intense source of wavelengths available in the x-ray and ultraviolet region.
20 Synchrotron radiation allows analysis of smaller quantities of sample that would be difficult to analyze using other sources of x-ray radiation.

One location for research using synchrotron radiation is the Stanford Synchrotron Radiation
25 Laboratory (SSRL), which is funded by the Department of Energy as a national user facility. Another location is Argonne National Laboratory, which is available to outside users on a fee basis.

30 Synchrotron radiation may be used to study structural details of solid samples with a resolution not practically attainable using

traditional x-ray instrumentation. This may enable differentiation between different polymorphic forms or compounds that is not attainable with other x-ray radiation sources.

5 Preferably, the present methods comprise generating more than one form such that a distribution of forms is obtained.

However, by generating solid forms in receptacles defining capillary spaces, one may
10 favor the formation of a variety of solid forms and increase the likelihood of generating all or a high percentage of possible forms.

The present methods can significantly assist in the identification of the form of a compound
15 or a mixture that is most stable or has other properties that make it preferable over other forms. For example, the present methods can be used as part of a screening method and can improve the likelihood of identifying a form
20 having biological activity such as better bioavailability, solubility, or adsorption characteristics. In some cases, an identified form may have better activity as an active agent.

After the sample is placed in a receptacle,
25 the receptacle may be centrifuged. Centrifugation may be employed for a variety of reasons. First, centrifuging may assist evaporation or concentrate solid or semisolid material at one end of a capillary space. This
30 has advantages in connection with in-situ analysis, in that the generated form will be located at a consistent place in the receptacle.

Also or alternatively, centrifuging may be used to provide additional environmental variation, which is desirable in a screening method.

5 **Example 1**

Portions of a sample of 4-(6-methoxy-2-naphthyl)-butan-2-one (Compound A) were dissolved in various solvents (acetone, acetonitrile, aqueous ethanol, ethanol, ethyl acetate, 10 tetrahydrofuran, toluene, benzene, chloroform, methyl ethyl ketone, methanol, butyl acetate, methylene chloride, hexane, aqueous tetrahydrofuran, aqueous methanol, aqueous acetone, and aqueous acetonitrile) to make 15 saturated solutions ranging in concentration from 5-50 mg/ml depending on the solvent. The solutions were filtered through 0.2 μ m nylon syringe filters into automatic pipettes. Aliquots (ranging from 5-25 microliters) of the 20 solutions were introduced into 40 glass capillaries (thin-walled, both ends open, half 0.7 mm inside diameter, half 1.0 mm inside diameter). For some of the capillary tubes, the original saturated solution was heated and more 25 4-(6-methoxy-2-naphthyl)-butan-2-one was added until the concentration was twice that of the saturation concentration. This supersaturated solution was then used.

The capillaries were rotated about their 30 center point at room temperature and solvent was allowed to evaporate until solid or semisolid material was visible by eye.

5 The resulting capillaries containing solid or semisolid material were analyzed by laboratory x-ray powder diffraction in the capillary tubes without isolation of material using an INEL XRG 3000 diffractometer. Analysis of the x-ray diffraction data showed four different x-ray powder patterns: the original crystalline form reported in the literature, two new crystalline powder patterns, and one amorphous pattern.

10 These four different x-ray diffraction patterns are indicative of four different solid forms. A comparative study of 4-(6-methoxy-2-naphthyl)-butan-2-one using 80 traditional screening conditions (including crystallization in vials, varying solvents, varying conditions including fast evaporation, slow cooling, and crash cooling) showed only one new diffraction pattern.

Example 2

20 Portions of a sample of sulfathiazole (compound B) are dissolved in various solvents (aqueous ethanol, acetonitrile, ethanol, methanol, aqueous methanol, methylene chloride, acetone, hexane, dioxane) to make saturated solutions ranging in concentration from 5-50 mg/ml depending on the solvent. The solutions are filtered through 0.2 μ m nylon syringe filters into automatic pipettes. Aliquots (ranging from 5-25 microliters) of the solutions are introduced

25 into 100 glass capillaries (thin-walled, single closed end, 0.7 mm inside diameter) and spun in a centrifuge to move the solution to the bottom of

the capillary tube. For some of the capillary tubes, the original saturated solution is heated and more Compound B is added until the concentration is twice that of the saturation
5 concentration. This supersaturated solution is then used.

The capillaries are placed in a variety of environments and solvent is allowed to evaporate until solid or semisolid material is visible by
10 eye. Environments include 60°C oven, 4°C freezer, ambient temperature, storage with closed end up, storage with closed end down, and spinning of the capillaries.

It is expected that the resulting
15 capillaries containing solid or semisolid material can be analyzed by laboratory x-ray powder diffraction in the capillary tubes without isolation of material using an INEL XRG 3000 diffractometer. Analysis of the x-ray
20 diffraction data would show whether different forms were present, including forms in addition to the known forms. Different x-ray diffraction patterns are indicative of different forms. A comparative study of sulfathiazole using 60
25 traditional screening conditions (crystallization in vials, varying conditions including fast evaporation, slow cooling, and crash cooling) would be expected to identify fewer different x-ray powder diffraction patterns.

30

Example 3

Portions of a sample of a polymorphic compound (compound C) are dissolved in various solvents (aqueous ethanol, methylene chloride, ethanol, toluene, dimethylformamide, acetone, water, butanol, methanol, acetonitrile, methylethylketone, hexane, dioxane, and ethyl acetate) to make solutions ranging in concentration from 5-50 mg/ml depending on the solvent. The solutions are filtered through 0.2 μ m nylon syringe filters into automatic pipettes. Aliquots (ranging from 5-25 microliters) of the solutions are introduced into 200 glass capillaries (thin-walled, single closed end, 0.7 mm inside diameter) and spun in a centrifuge to move the solution to the bottom of the capillary tube, which facilitates in situ analysis.

Aliquots (ranging from 5-25 microliters) of the solutions are also introduced into 100 double open-ended glass capillaries (thin-walled, double open ends, 1.0 mm inside diameter).

The capillaries are placed in a variety of environments and solvent is allowed to evaporate until solid or semisolid material it was visible by eye. Environments include 60°C oven, 4°C freezer, ambient temperature, storage with closed end up, and storage with closed end down. Some of the capillaries are stored under centrifugation at 40°C and ambient pressure while the solvent evaporation took place. The 100

open-ended capillaries are rotated about their center point during solvent evaporation.

The resulting capillary tubes containing solid or semisolid material can be analyzed by
5 synchrotron x-ray powder diffraction. It is expected that this in situ analysis of the x-ray diffraction data would show different patterns corresponding to different forms, and that more forms would be observed than if the forms were
10 generated by a traditional screening method. Different x-ray diffraction patterns are indicative of different forms of the compound.

A comparative study using traditional screening techniques to prepare different forms
15 of the same compound would be expected to identify fewer different x-ray diffraction patterns.

Example 4

Solutions of an organic drug sample (compound D) are prepared in a similar way as those in Example 2. Aliquots (15-20 microliters each) of the various solutions are placed in two glass, thin-walled 96-well plates with well dimensions of approximately 2 mm x 2 mm x 8 mm. The solutions are evaporated by placing one plate in a SpeedVac centrifugal evaporator at 30°C and 25 mm Hg vacuum and one in a SpeedVac centrifugal evaporator at 50°C and 100 mm Hg vacuum. The different evaporation conditions provide different evaporation rates and other environmental variations. The resulting solid and semisolid residues are analyzed in situ by transmission x-ray powder diffraction. Analysis of the x-ray data would be expected to show distinct powder patterns for the different forms generated.

20

Example 5

Following a procedure having the same steps as Example 1, forms are generated. After solutions in capillary tubes evaporate to leave
5 solid or semisolid residue, the capillary tubes are cut to a 2 cm length containing the bulk of the residue and then crushed and analyzed by infrared (IR) spectroscopy. Analysis of the IR data would be expected to indicate presence of
10 several different forms, that is, several distinguishable IR patterns. Different IR patterns are indicative of different forms.